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Acids react with bases to produce salts and water. 5. Acids conduct electric current. Binary Acid. An acid that contains only two different elements: hydrogen and one of the more electronegative elements. Oxyacid. An acid that is a compound of hydrogen, oxygen, and a third element, usually a nonmetal.

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Define a Lewis acid. Can H+ function as a Lewis acid? Explain your answer. A Lewis acid is a species that accepts an electron pair to form a covalent bond. H+ is a Lewis acid because it can accept an electron pair from a base. For example, 3. Identify the Brl1lnsted-Lowry acid and the Brl1lnsted-Lowry base on the reactant side of each of the

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