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Chemistry: Chapter 5 Review. Describe... 1) Dalton ' s model – solid sphere, made up of only 1 type of matter, no space. 2) Schrodinger ' s model – based on probability of finding an e⁻; electrons are found in orbitals (look like electron clouds), which are in sublevels (s,p,d,f,g), which are in energy levels.

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CHAPTER 5 REVIEW The Periodic Law SECTION

5-3 SHORT ANSWER Answer the following

questions in the space provided. 1. When an electron is added to a neutral atom, energy is . (a) always absorbed (c) either absorbed or released (b) always released (d) burned away 2.

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Section 3

CHAPTER 5 REVIEW The Periodic Law MIXED REVIEW

SHORT ANSWER Answer the following questions in the space provided. 1. Consider the neutral atom with 53 protons and 74 neutrons to answer the following questions. 53 a. What is its atomic number? 127 b. What is its mass number? atomic number c. Is the element's position in a modern periodic table determined by

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