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Chemistry: Chapter 5 Review. Describe.... 1) Dalton's model – solid sphere, made up of only 1 type of matter, no space. 2) Schrodinger's model – based on probability of finding an e⁻; electrons are found in orbitals (look like electron clouds), which are in sublevels (s,p,d,f,g), which are in energy levels.

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5 The Periodic Law

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CHAPTER 5 REVIEW The Periodic Law SECTION 5-3 SHORT ANSWER Answer the following questions in the space provided. 1. When an electron is added to a neutral atom, energy is . (a) always absorbed (c) either absorbed or released (b) always released (d) burned away 2.

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Chapter 6 Review

How to Write the Electron Configuration for an Element in Each Block

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(STEREOCHEMISTRY)

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Integration, IB, AP, & ABC Chapter 4:5 Notes Part II Chapter 5 -

Stereochemistry: Part 1 of 6

CHAPTER 5 REVIEW The Periodic Law MIXED REVIEW SHORT

ANSWER Answer the following questions in the space provided. 1. Consider

the neutral atom with 53 protons and 74 neutrons to answer the following

questions. 53 a. What is its atomic number? 127 b. What is its mass number?

atomic number c. Is the element 's position in a modern periodic table

determined by

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