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Modern Chemistry 68 Chapter Test
Name Class Date Chapter Test A,
continued ____ 21. Metal X replaces
the ions of metal Y from solution, but it
cannot replace the ions of metal Z
from solution. The order these metals
should have in the activity series
(from top to bottom) is a. x, y, z b. z,
x, y

Chemistry Course: Modern Chemistry Chapter
8 Test

Modern Chemistry 91 Chapter Test Name
Class Date Chapter Test B, continued 27.
Distinguish between ionic crystals and metallic
crystals. PART V Write the answers to the
following questions on the line to the left, and
show your work in the space provided. The
molar enthalpy of fusion for water is 6.008

kJ/mol. 28.

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Modern Chemistry 111 Chapter
Test Name Class Date Chapter
Test B, continued 18.

Dissolution processes with
negative enthalpies of
solution are processes. 19.
is the solution process with
water as the solvent. 20.
Liquid solutes and solvents
that are not soluble in each
other are. 21. A solute
molecule that is surrounded
by solvent molecules is ...

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Chemistry. CH 1 Vocabulary-New. CH 1
Mixed Questions. ... CH 5 Self Test. CH 5
Evaluation . CH 5 Test . CH 5 Periodic Trends
Practice Test .

Assessment Chapter Test A

Holt McDougal Modern Chemistry 1 Chapter

Test Assessment Chapter Test A Chapter: The
Periodic Law Use the periodic table below to
answer the questions in this Chapter Test. In
the space provided, write the letter of the term
or phrase that best completes each statement or
best answers each question. _____ 1.

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Flashcards and Study ...

MODERN CHEMISTRY The atomic
number of nickel-60 is 28. How many neutrons
does this iso- tope have? Carbon-14 has 8
neutrons. What is the atomic number of
carbon-14? 11 CHAPTER 3 TEST HR W
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McDougal Modern Chemistry's States of

Matter chapter with which you need help.
Find the corresponding video lessons within
this companion course chapter.

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and Study Sets ...

Modern Chemistry 84 Chapter Test Chapter:
States of Matter In the space provided, write
the letter of the term or phrase that best
completes each statement or best answers each
question. _____ 1. The kinetic-molecular
theory of gases assumes that the particles of an
ideal gas are separated by great distances. This
implies that the gas

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rapport, there can be little doubt that
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at once in her brain where generally the
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actions are heavily affected by this.

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CHAPTER 3 REVIEW Atoms: The
Building Blocks of Matter MIXED

REVIEW SHORT ANSWER Answer the
following questions in the space provided. 1.
The element boron, B, has an atomic mass

of 10.81 amu according to the periodic table. decomposition of a substance by an electric current is called . 19. A(n) orders the elements by the ease with which they undergo certain chemical reactions. 20.

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Modern Chemistry 71 Chapter Test Name Class Date Chapter Test B, continued 17. A substance combines with oxygen, releasing a large amount of energy as heat and light, in a(n) . 18. The

Assessment Chapter Test B - Wag & Paws

Modern Chemistry 47 Chapter Test Name Class Date Chapter Test A, continued ____ 7. The melting points of ionic compounds are higher than the melting points of molecular compounds because a. ionic substances tend to vaporize at room temperature. b. ionic substances are brittle. c. attractive forces between ions are greater than the attractive ...

Assessment Chapter Test A

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