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Name Date Class 10—2 Practice Problems 1. Find the mass of 0.89 mol of CaC_2 .

Determine the number of atoms that are in 0.58 mol of Se. q A bottle of PbSO_4 contains 158.1 g of the How many moles of barium nitrate

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CHAPTER 3 REVIEW Atoms: The Building Blocks of Matter MIXED REVIEW SHORT ANSWER Answer the following questions in the space provided. 1. The element boron, B, has an atomic mass of 10.81 amu according to the periodic table.

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the rates of effusion of gases at the same temperature and pressure are inversely proportional to the square roots of their molar masses Rate of effusion of A / B = $\sqrt{\frac{M_b}{M_a}}$

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