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3 Atoms: The Building Blocks of Matter

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Modern Chemistry Study Guide Answers

SHORT ANSWER Answer the following questions in the space provided. 1. Explain the difference between the mass number and the atomic number of a nuclide. Mass number is the total number of protons and neutrons in the nucleus of an isotope. Atomic number is the total number of protons in the nucleus of each atom of an element. 2.

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SHORT ANSWER Answer the following questions in the space

provided. 1. Classify each of the following as a homogeneous or

heterogeneous substance. a. sugar d. plastic wrap b. iron filings e.

cement sidewalk c. granola bar 2. For each type of investigation,

select the most appropriate branch of chemistry from the following

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