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Date Chapter Test B, continued 18. When a
strong acid is titrated with a weak base, the pH of
the solution at the equivalence point is than 7. 19.
A is a highly purified solid used to check the
concentration of a standard solution. 20. A 1 M

solution of NaOH will have a pH that is
Modern Chemistry Chapter Atoms Test Answers
Modern Chemistry 20 Chapter Test Name Class
Date Chapter Test A, continued ____13. To
calculate the number of atoms present in 2.0 mol
of an element, you would a. add Avogadro's
number of atoms per mole to 2.0 mole. b. subtract
Avogadro's number of atoms per mole from 2.0
mole. c.

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hydrogen in MgH₂. 0 c. sulfur.
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Modern Chemistry 88 Chapter Test Chapter:
States of Matter PART I In the space
provided, write the letter of the term or phrase
that best completes each statement or best
answers each question. _____ 1. According to
the kinetic-molecular theory, particles of an

ideal gas a. attract each other but do not collide. b. repel each other and collide.
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Modern Chemistry Test B Answer
Modern Chemistry 35 Chapter Test Name Class Date Chapter Test B, continued Write the orbital notation for the following elements in the space provided. 37. lithium, atomic number 3 38. carbon, atomic number 6 39. neon, atomic number 10 PART VI Write the answers to the questions on the line

to the left, and show your work in the space provided.
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Holt McDougal Modern Chemistry 3 Chapter Test Chapter Test B, continued 16 Modern chemistry chapter 3 test b answers. The measure of the ability of an atom in a chemical compound to attract electrons from another atom in the compound is called _____. 17. The energy required to remove one electron from an atom is called its _____. 18.
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Modern Chemistry 32 Chapter Test Name Class Date Chapter Test B, continued ____ 6. kindly help me to find out the answer keys. Significant emphasis is placed on theoretical aspects of chemistry, in-depth laboratory experiences, and problem solving. 7 I usually finish at four in the morning, (2).
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Assessment Chapter Test B - Wag & Paws Modern Chemistry 33 Chapter Test Name Class Date Chapter Test B, continued 15. The energy state of an atom is called its ground state. 16. The number of waves that pass a point in one second

is called. 17. When an electron drops from a higher-energy state to a lower-energy state, a(n) spectrum is produced. 18.

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Chapter 6 Modern Chemistry Test Answers

PROBLEMS Write the answers to the questions on the line to the left, and show your work in the space provided. 25. 26. MODERN CHEMISTRY The atomic number of nickel-60 is 28. How many neutrons does this iso- tope have?

Carbon-14 has 8 neutrons. What is the atomic number of carbon-14? 11 CHAPTER 3 TEST Modern Chemistry Chapter 7 Test B

Answers

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