

# Mole Ratio Answer Key

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The following six mole ratios can be written for the ammonia forming reaction above. (12.2.3) 1 mol N<sub>2</sub> 3 mol H<sub>2</sub> or 3 mol H<sub>2</sub> 1 mol N<sub>2</sub> 1 mol N<sub>2</sub> 2 mol NH<sub>3</sub> or 2 mol NH<sub>3</sub> 1 mol N<sub>2</sub> 3 mol H<sub>2</sub> 2 mol NH<sub>3</sub> or 2 mol NH<sub>3</sub> 3 mol H<sub>2</sub>. In a mole ratio problem, the given substance, expressed in moles, is written first.

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Mole ratios are used as conversion factors between products and reactants in stoichiometry calculations. For example, in the reaction .

2H<sub>2</sub>(g) + O<sub>2</sub>(g) → 2H<sub>2</sub>O(g) The mole ratio between O<sub>2</sub> and H<sub>2</sub>O is #(1

mol O<sub>2</sub>)/(2 mol H<sub>2</sub>O)#. The mole ratio between H<sub>2</sub> and H<sub>2</sub>O is #(2 mol H<sub>2</sub>)/(2 mol H<sub>2</sub>O)#. Example:

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## Chemical Quantities (Stoichiometry)

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Stoichiometry Mole to Mole Ratio (Tagalog-  
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(4) two moles of NH<sub>3</sub> are produced from one mole of N<sub>2</sub> .  
this can be written as the ratio (2 NH<sub>3</sub>)/(1 N<sub>2</sub>) (5) three  
moles of H<sub>2</sub> produce two moles of NH<sub>3</sub>. this can be  
written as the ratio (3 H<sub>2</sub>)/(2 NH<sub>3</sub>) Again, we can write  
the inverse of this: (6) two moles of NH<sub>3</sub> are produced  
from three moles of H<sub>2</sub> . this can be written as the ratio  
(2 NH<sub>3</sub>)/(3 H<sub>2</sub> ...

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#### Chapter 3: Stoichiometry - Mrs. Penney

Mole Ratio Worksheet. 1) Balance this equation:  $N_2 + H_2 \rightarrow NH_3$ , write the following molar ratios:  $N_2 / H_2$   $N_2 / NH_3$   $H_2 / NH_3$ . 2) Balance this equation:  $H_2 + S \rightarrow H_2S$ , write the following molar ratios:  $H_2 / H_2S$   $H_2 / S$   $H_2S / S$ . 3) Write and balance the equation for the synthesis of water. Then answer the following questions.

#### Eleventh grade Lesson Mole Ratios | BetterLesson

PDF Mole Conversion Worksheet Answer Key in this set (33) How many moles in 28 grams of  $CO$  ? Molar Mass (grams) of  $CO$  :  $1C = 1 \times 12.01g = 12.01g$  Mole Conversion Practice Answer Key Diagram | Quizlet Mole to Grams, Grams to Moles Conversions Worksheet What are the molecular weights of the following compounds? (all masses must be to Page 5/29

#### Mole Ratio Wkst Answers

I also know that balanced chemical equations are necessary for the proper use of mole ratios. Activator: I

ask a student to show their work or share their coefficients. They should be:  $2 H_2 O + O_2 \rightarrow 2 H_2 O$   
 $2. 2 NaBr + CaF_2 \rightarrow 2 NaF + CaBr_2$ .  $HNO_3 + NaHCO_3 \rightarrow NaNO_3 + H_2 O + CO_2$ .

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