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Avogadro ' s principle – says that equal volumes of gases at the same temp. and pressure contain equal numbers of molecules.

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that are gases near room temperature, except the

noble gases, normally exist as diatomic molecules. 334 CHAPTER 11 FIGURE 11-1 At the same temperature and pressure, balloons of equal volume have equal numbers of molecules, regardless of which gas they contain. Hydrogen molecule 1 mol H₂ at STP = 22.4 L Oxygen molecule 1 mol O₂ at STP = 22.4 L Carbon dioxide

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