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[Writing Formulas with Polyatomic Ions 05 CTR ch09 7/9/04 3:29 PM Page 211 NAMING IONS 9 Section Review Objectives ... or the naming system. 10. Ions containing](#)

more than one atom are called ions. 11. The names of most common polyatomic ions end in either 12. [Filename: chsr91.pdf] - Read File Online - Report Abuse

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[Answers To Section 9 Review *Naming Ions Section Review 7.1 Part A Completion . valence electrons 2. group electron dot structures 3. octet rule 4. 5. cations 6. amons Halide ions 8. 9. gain charges 10.*](#)

15. AT 16. NT 22. a 23. c True/False 13.
ST 14. NT Matching 20. g 21. f Part B 11.
NT 12. AT Part C 18. 19. b d e
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The cation name is placed first when naming ionic
compounds. AT. Binary Compounds. B. ionic
compounds composed of two elements.
Monatomic ions. A. ions that consist of a single
atom. Polyatomic Ion. D. Ions that consist of more
than one atom. Transition Metals. ... 9.1 Naming
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Chapter 9 Chemical Names And Formulas Section
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an ion. It also defines polyatomic ion and
gives the names and formulas for the most
common polyatomic ions. Monatomic Ions
(pages 253 – 256) 1. What are monatomic
ions? 2. How is the ionic charge of a Group
1A, 2A, or 3A ion determined? 3. How is
the ionic charge of a Group 5A, 6A, or 7A
ion determined? 4. Circle the letter of the
type of element that often has more than
one common ionic charge.

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9.2 Naming and Writing formulas for ionic
compounds ...

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9.4 – Naming and Writing Formulas for

Acids and Bases. An acid is a compound that
produces H⁺ ions when it dissolves in water.
The formula for an acid normally starts with
and H.

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Analyze Identify the relevant concepts. You
can ...

Ions that consist of a single atom are called
ions. 1. Metallic elements tend to electrons.
Group 1A ions have a 2. charge, whereas
Group 2A metals form ions with a 3.
charge, and Group 3A metals form ions
with a charge. 4. The charge of a Group A
nonmetal ion is determined by 5.
subtracting from the group number. For
example, the 6.