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Objective NCERT Based Chapterwise Topicwise Solutions For 11th And 12th Class with Solved Papers (2005 -2023) with Notes for NEET-AIIMS Exam 2024 - Chemistry Disha Publications

NCERT Exemplar Chemistry Problems - Solutions (Class 11) is a comprehensive book for students of standard XI studying in schools affiliated to the Central Board of Secondary Education. The book comprises chapters on structure of atom, classification of elements and periodicity of properties, chemical bonding and molecular structure, states of matter, equilibrium, redox reactions and hydrocarbons. In addition, the book consists of several solved examples for thorough revision and final practice.

NCERT Solutions Mathematics Class 11th Oswaal Books and Learning Private Limited An Excellent Book in Accordance with the latest syllabus for Class-11 Prescribed by CBSE/NCERT and Adopted by Various State Education Boards. (A) Basic Laboratory Techniques – 1. To cut a glass tube or glass rod, 2. To bend the glass rod at an angle, 3. To draw a glass jet from a glass tube, 4. To bore a cork and fit a glass tube into it. (B) Characterisation and Purification of Chemical Substances- 1. To determine the melting point of the given unknown organic compound and its identification (simple laboratory technique), 2. To determine the boiling point of a given liquid when available in small quantity (simple laboratory method), 3. To prepare crystals of pure potash alum [K2SO4.Al2(SO4)3.24H2O] from the given impure sample, 4. To prepare the pure crystals of copper sulphate from the given crude sample, 5. To prepare pure crystals of benzoic acid from a given impure sample. (C) Measurement of pH Values 1. To determine the pH value of vegetable juices, fruit juices, tap water and washing soda by using universal pH paper, 2. To determine and compare the pH values of solutions of strong acid (HCI) and weak acid (CH3COOH) of same concentration, 3. To study the pH change in the titration of strong base Vs. strong acid by using universal indicator paper, 4. To study the pH change by common ion (CH3COO – ion) i case of weak acid (CH3COOH), 5. To determine the change in pH value of weak base (NH4OH) in presence of a common ion (NH4+), (D) Chemical Equilibrium 1. To study the

shift in equilibrium between ferric ions and thiocyanate ions by changing the concentrations of either of the ions, 2. To study the shift in equilibrium between [Co(H2O)6]2+ and Cl - ions by changing the concentrations of either of the ions, (E) Quantitative Analysis 1. To prepare M/10 oxalic acid solution by direct weighing method, 2. To prepare M/10 solution of sodium carbonate by direct weighing method, 3.To determine the strength of given solution of sodium hydroxide by titrating it against N/10 or M/20 solution of oxalic acid, 4. To determine the strength of a given solution of hydrochloric acid by titrating it against a standard N/10 or M/20 sodium carbonate solution, (F) Qualitative Analysis 1. Analysis of Anions, 2. Analysis of Cations (G) Detection of Elements in Organic Compounds 1.To detect the presence of nitrogen, sulphur and halogens in a given organic compound by Lassaigne 's test, 2. To detect the presence of nitrogen, sulphur and halogens in the given organic compound sample number by Lassaigne 's test INVESTIGATORY PROJECTS (A) Checking of Bacterial Contamination in Water 1. To check the bacterial contamination in drinking water by testing sulphide ions (B) Methods of Water Purification 1.To purify water from suspended impurities by using sedimentation, 2. To purify water by boiling, 3. To purify water by distillation method, 4. To purify water by reverse osmosis technique. 5. To purify water by GAC method, 6. To purify water by bleach treatment, 7. To purify water by oxidising agent, 8. To purify water by ozone treatment method. (C) Water Analysis 1. To test the hardness of different water samples. (D) Foaming Capacity of Various Soaps 1.To compare the foaming capacity of different washing soaps, 2. To study the effect of addition of sodium carbonate on foaming capacity of washing soap (E) Tea Analysis 1. To study the acidity of different samples of tea leaves (tea) by using pH paper (F) Analysis of Fruits and Vegetable Juices 1. To analyse the fruit and vegetable juices for the constituent present in them (G) Rate of Evaporation 1. To study the rate of evaporation of different liquids (H) Effect of Acids and Bases on Tensile Strength of Fibres 1.To compare the tensile strength of natural fibres and synthetic fibres, 2. To study the effect of acids and bases on tensile strength of different fibres. Log & Antilog Table

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NCERT Textbooks play the most vital role in developing student's understanding and knowledge about a subject and the concepts or topics covered under a particular subject. Keeping in mind this immense importance and significance of the NCERT Textbooks in mind, Arihant has come up with a unique book Environmental Chemistry I. Appendix II. Log-antilog Table containing Questions-Answers of NCERT Textbook based questions. This book containing solutions to NCERT Textbook questions has been designed for the students studying in Class XI following the NCERT Textbook for Biology. The present book has been divided into 22 Chapters namely Biological Classification, Plant Kingdom, Animal Kingdom, Biomolecules, Mineral Nutrition, Respiration in Plants, Digestion & Absorption, Anatomy of Flowering Plants, Cell Cycle & Cell Division, Respiration in Plants, Body Fluids & Circulation, Morphology of Flowering Plants, Locomotion & Movement, etc. covering the syllabi of Biology for Class XI. This book has been worked out with an aim of overall development of the students in such a way that it will help students define the way how to write the answers of the textbook based questions. The book covers selected NCERT Exemplar Problems which will help the students understand the type of questions and answers to be expected in the Class XI Biology Examination. Also each chapter in the book begins with a summary of the chapter which will help in effective understanding of the theme of the chapter and to make sure that the students will be able to answer all popular questions concerned to a particular chapter whether it is Long Answer Type or Short Answer Type Question. For the overall benefit of students the book has been designed in such a way that it not only gives solutions to all the exercises but also gives detailed explanations which will help the students in learning the concepts and will enhance their thinking and learning abilities. As the book has been designed strictly according to the NCERT Textbook of Biology for Class XI and contains simplified text material in the form of class room notes and answers to all the questions in lucid language, it for sure will help the Class XI students in an effective way for Biology.

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Content: 1. Some Basic Concepts of Chemistry, 2. Structure of Atom, 3. Classification of Elements and

Periodicity in Properties, 4. Chemical Bonding and Molecular Structure, 5. States of Matter, 6. Thermodynamics, 7. Equilibrium, 8. Redox Reactions, 9. Hydrogen, 10. s-Block Elements 11. p-Block Elements, 12. Organic Chemistry—Some Basic Principles and Techniques 13. Hydrocarbons 14.

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NCERT Textbooks play the most vital role in developing student's understanding and knowledge about a subject and the concepts or topics covered under a particular subject. Keeping in mind this immense importance and significance of the NCERT Textbooks in mind, Arihant has come up with a unique book containing Questions-Answers of NCERT Textbook based questions. This book containing solutions to NCERT Textbook questions has been designed for the students studying in Class XI following the NCERT Textbook for Physics. The present book has been divided into 15 Chapters namely Physical World, Motion in a Plane, Laws of Motion, Work, Energy & Power, Gravitation, Thermodynamics, Kinetic Theory, Oscillations, Waves, Motion in a Straight Line, Thermal Properties of Matter, Mechanical Properties of Solids, etc covering the syllabi of Physics for Class XI. This book has been worked out with an aim of overall development of the students in such a way that it will help students define the way how to write the answers of the Physics textbook based questions. The book covers selected NCERT Exemplar Problems which will help the students understand the type of questions and answers to be expected in the Class XI Physics Examination. Also each chapter in the book begins with a summary of the chapter which will help in effective understanding of the theme of the chapter and to make sure that the students will be able to answer all popular questions concerned to a particular chapter whether it is Long Answer Type or Short Answer Type Question. For the overall benefit of students the book has been designed in such a way that it not only gives solutions to all the exercises but also gives detailed explanations which will help the students in learning the concepts and will enhance their thinking and learning abilities. As the book has been designed strictly according to the NCERT Textbook of Physics for Class XI and contains simplified text material in the form of class room notes and answers to all the questions in lucid language, it for sure will help the Class XI students in an effective way for Physics.

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Syllabus: Unit I: Some Basic Concepts of Chemistry, Unit II: Structure of Atom, Unit III: Classification of Elements and Periodicity in Properties, Unit IV: Chemical Bonding and Molecular Structure, Unit V: States of Matter: Gases and Liquids, Unit VI: Chemical Thermodynamics, Unit VII : Equilibrium, Unit VIII : Redox Reactions, Unit IX : Hydrogen, Unit X : s-Block Elements (Alkali and Alkaline earth metals) Group 1 and Group 2 Elements, Unit XI: Some p-Block Elements General Introduction to p-Block Elements, Unit XII: Organic Chemistry—Some Basic Principles and Techniques, Unit XIII: Hydrocarbons Classification of Hydrocarbons, Unit XIV: Environmental Chemistry Content : 1. Some Basic Concepts of Chemistry, 2. Structure of Atom, 3. Classification of Elements and Periodicity in Properties, 4. Chemical Bonding and Molecular Structure, 5. States of Matter, 6. Thermodynamics, 7. Equilibrium, 8. Redox Reactions, 9. Hydrogen, 10. s-Block Elements 11. p-Block Elements, 12. Organic Chemistry—Some Basic Principles and Techniques 13. Hydrocarbons 14. Environmental Chemistry I. Appendix II. Log-antilog Table

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