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Types of Solution. Liquid solutions, such as sugar in water, are the most common kind, but there are also solutions that are gases or solids. Any state of matter (solid, liquid, or gas) can act both as a solute or as a solvent during the formation

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solids:  
Properties of  
solids (i)  
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definite  
shape and  
distinct  
boundaries.  
(ii) Solids  
have fixed  
volume. (iii)

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They have negligible compressibility. (iv) They are rigid (their shape cannot be changed). (v) Their intermolecular force of attraction is maximum. (vi) The kinetic energy of its particles is minimum.

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