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Organic chemistry was traditionally defined as the study of substances isolated from living systems. While organic chemistry opens the door to the understanding of living systems including but not limited to protein chemistry and pharmaceuticals, organic chemistry is also the basis of materials chemistry which has given us such commercial materials as fiberglass, plastic, lasers, and computers to name a few.

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The goal is not to understand every single thing about this organic chemistry mechanism in great detail right now. Our goal right now is just to appreciate how nucleophiles and electrophiles are used in organic chemistry mechanisms and start to get a feeling for how these curved arrows show the movement or flow of electrons.

Organic Chemistry Principles And Mechanisms

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Organic chemists use a curved arrow (\curvearrowright) to show the movement of an electron pair. A sigma(σ) bond is formed between atoms by the overlap of two atomic orbitals along the line that connects the atoms. Carbon uses sp^3 -hybridized orbitals to form four such bonds. These bonds are directed from the carbon nucleus toward the corners of a tetrahedron.

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Structure and Mechanism in Organic Chemistry. C. K. Ingold. Ithaca, N. Y.: Cornell Univ. Press, 1953. 828 pp. Illus. \$9.75

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