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mass of an element is the  
weighted average mass of  
the isotopes of the element.  
(1) It is measured on a scale  
on which a carbon-12 atom  
has a mass of exactly 12  
units.(1). b)The total mass  
of 100 atoms =  $(75 \times 35) +$   
 $(25 \times 37) = 3550$  (1) The  
average mass of 1 atom =  
 $3550/100 = 35.5$  (1)

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