
Periodic Properties 23 Experiment Answers

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EXPERIMENT 4 –
The Periodic
Table - Atoms
and Elements
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Solutions 2 Chapter
9 - The Mole
Concept 9.1 Since
a sulfur atom ($A_r = 32$) weighs twice as
much as an oxygen
atom ($A_r = 16$), 32
g of sulfur atoms
and 16 g of oxygen
atoms both contain
the same number
of atoms.

*Periodic Properties
23 Experiment
Answers*
For this part of the
experiment you will
make a graph of a
periodic property
vs. atomic number.
The property
chosen is atomic
radius, given in pm
(1 pm = 10^{-12} m).
The atomic radius,
 r , is related to the
volume of the atom
by the formula of a
sphere, $V = \frac{4}{3} r^3$.
In Table 1 below,
the atomic radii of
the first 54
elements are given.
Graphing periodic
trends worksheet
answer key
Also, magnesium
and calcium are in
different periods on
the periodic table,
which may indicate
that they react
differently. 2.) List
the four metals
from most reactive

to least reactive.
Use data from your lab to support your answer. Zinc (most reactive)
Magnesium
Calcium Tin (least reactive) (refer to tables for evidence)
3.)
Experiment:
Periodic Properties of the Elements
Experiment
Answers Periodic Properties 23 Experiment
Answers Your lab report will include all of the observations that you made and the answers to the lab questions. It is critical that you wear lab aprons and goggles for the entire experiment!!
Purpose: To determine trends in reactivity in the periodic table by

examining the Alkali Metals and ...
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Answers
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Periodic Trends in Reactivity - Bonnie May's Digital Portfolio
The Periodic Table was originally devised by Dmitri Mendeleev in

Russia in 1869 as a result of the careful observations he made of the periodic properties of the known elements. He fortunately had the atomic masses available and was, therefore, able to arrange the elements according to relative mass and according to chemical and physical properties.

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Nicole Fruehauf-20
20-09-29-16-38-44
Subject: Periodic Properties Lab

Answer Keywords
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Answers
Lab - Graphing Periodic Properties - Name Class Date
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Mar 24, 2014 - ...all
things were created
by Him and for
Him. Colossians
1:16 These
experiments and
demonstrations
have been proven
in our
homeschooling of
ages 6-13. If we
didn't have an
experiment for
each element, we
supported our
study with a related
article, a household
object to represent
the element in our
"element box" or a
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Periodicvideos.com
. [23.1: General](#)

[Properties of
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Chemistry ...](#)
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Ionization
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Atomic Radius,
Ionization
Energy, and
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Table: Crash
Course
Chemistry #4
Periodic Table |
Previous 25
Years Questions
| Complete In 60
Min. | NEET |
JEE | Vishal Sir
Trick to find
group number
and period**

<u>number in</u>	<i>Memorise The</i>	Table AP IIIT
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<u>table/periodic</u>	<i>Song SCIENCE</i>	SCIENCE
<u>table tricks/class</u>	<i>SONGS The</i>	PRACTICE
<u>12 chem</u>	<i>Origin of the</i>	TEST - 39
<u>(new) - Periodic</u>	<i>Elements Which</i>	CLACIFICATION
<u>Table of Videos</u>	<i>Came First—The</i>	OF ELEMENTS
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<u>Periodic Table of</u>	<i>Egg? How Small</i>	TABLE AP
<u>Videos mcq</u>	<i>Is An Atom?</i>	Chemistry:
<u>#periodic</u>	<i>Spoiler: Very</i>	1.5-1.8 Atomic
<u>classification of</u>	<i>Small. They</i>	Structure,
<u>elements : Quiz</u>	<i>Might Be Giants -</i>	Electron
<u>:CBSE 10th</u>	<i>Meet the</i>	Configuration,
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<u>Periodic Table </u>	Lyrics—Harry	Elements, Metric
<u>ICSE Class 10</u>	Potter Song	System \u0026
<u>Chemistry</u>	(Riddle™)	<u>Unit Conversion</u>
<u>Umang</u>	Periodic Table of	<u>Matter</u>
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answers could go
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the solutions for
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answers