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Answer: C. 3.
An atom located in the third period of the periodic table would have an electron configuration between $1s^22s^22p^63s^1$ and $1s^22s^22p^63s^23p^6$.
The highest energy level is 3 as can be seen from the electron configurations.
6.
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In the late 1800's, Dmitri Medeleev devised the first periodic table based on _____. atomic mass In 1913, Henry G. J. Moseley arranged the elements by ____ rather than atomic mass.

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A section of the

periodic table showing metals and non-metals. The main groups are numbered from 1 to 7 going from left to right, and the last group on the right is group 0. The section in the ...

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How many columns
of elements does the
periodic table
contain? ___18___

7. What is another
name for a column
of elements?

___group or
family_____ 8.

What two group
numbers can be
used to designate
elements in the
second column of
the periodic table?

___group 2A or
group 2_____ 9.

How many rows of
elements does the
periodic table
contain ...