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Periodic Table - Ptable

the periodic table chapter 6 section 6.3 answer key, 6. How many columns of elements does the periodic table contain? ____18____ 7. What is another name for a column of elements? ____group or family____ 8. What two group numbers can be used to designate elements in the second column of the periodic table? ____group 2A or group 2____ 9. How many rows of elements does the periodic table contain ...

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A section of the periodic table showing metals and non-metals. The main groups are numbered from 1 to 7 going from left to right, and the last group on the right is group 0. The section in the ...

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In the late 1800's, Dmitri Medeleev devised the first periodic table based on _____. atomic mass In 1913, Henry G. J. Moseley arranged the elements by ____ rather than atomic mass.

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Answer: C. 3. An atom located in the third period of the periodic table would have an electron configuration between 1s22s22p63s1and 1s22s22p63s23p6. The highest energy level is 3 as can be seen from the electron configurations. 6.

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