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A zero, as acid solutions do not contain OH⁻ ions B 1.0×10^{-12} mol dm⁻³ C 0.01 mol dm⁻³ D 0.30 mol dm⁻³ E 3.9 Concentrated nitric and sulfuric acids react according to the equation: $\text{HNO}_3 + \text{H}_2\text{SO}_4 \rightarrow \text{H}_2\text{NO}_3 + \text{HSO}_4$ This reaction shows that: A sulfuric acid is a stronger acid than nitric acid B nitric acid is a stronger ...

A zero as acid solutions do not contain OH⁻ ions B 10^{-10} C 10^{-12} ...

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Worksheet 5.1 HYDROCARBONS, HALOGENOALKANES & ALCOHOLS ALCOHOLS © Philip Allan Updates TOPIC5 Structure and reactions

Questions (1) Name the following compounds: CH₃ ...

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