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1/2 3 2 4 3 2. 2N O (g) 4NO (g) + O (g) Rate = k N O CHCl (g) + Cl (g) CCl (g) + HCl(g) Rate = k[CHCl ] [Cl ] ?  
?. Answer: The rate of the reaction in Equation 14.9 is first order in N2O5and first order overall. The reaction in Equation 14.10 is first order in CHCl3and one-half order in Cl2.  
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