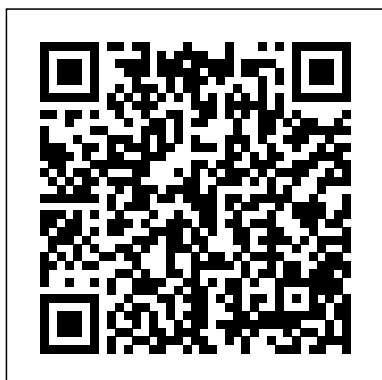

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percentage composition: 39,9% carbon 6,7% hydrogen 53,4% oxygen Determine the molecular formula of acetic acid if the molar mass of acetic acid is 60 g/mol-1. (6)

6.2.2 Give a reason why acetic acid is regarded as a monoprotic acid. (2)

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