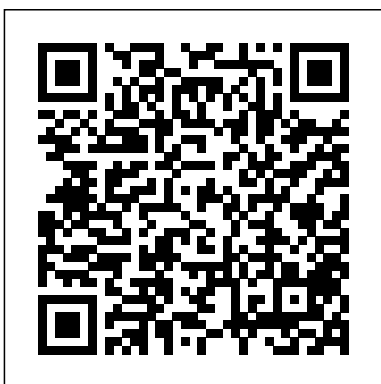


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Chemists combine all of the relationships seen in Models 1 and 2 into one law—the Ideal Gas Law. It is one equation that describes gas behavior and the relationship among all four variables, P, V, T, and n. In the Ideal Gas Law the proportionality constant is represented by the letter R (rather than the generic k). 24.

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one increases, so does the other =

Inversely Proportional As

one increases, the other decreases =

Thermodynamic

Observables Aspects of a system that we can observe, measure, or calculate in some way. Pressure – P kilopascals (kPa) Temperature – T Kelvin (K) Volume – V Liters (L) moles – n ...

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Volume = 1

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