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Studyguide for Principles of General Chemistry by Silberberg, Martin, ISBN 9780077491260 McGraw-Hill Science/Engineering/Math

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Silberberg's Principles of General Chemistry offers students the same authoritative topic coverage as its parent text, Chemistry: The Molecular Nature of Matter and Change. The Principles text allows for succinct coverage of content with minimal emphasis on pedagogic learning aids. This more streamlined approach to learning appeals to today's efficiency-minded, value-conscious instructors and students without sacrificing depth, clarity, or rigor.

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Principles of General Chemistry + Aleks for General Chemistry
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Chemistry: The Molecular Nature of Matter and Change Academic
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The seventh edition of General Chemistry continues the tradition of presenting only the material that is essential for a one-year general chemistry course. It strikes a balance between theory and application by incorporating real-world examples; helping students visualize the three-dimensional atomic and molecular structures that are the basis of chemical activity; and developing problem-solving and critical thinking skills. Although the seventh edition incorporates many impressive features, such as conceptual idea review, animations correlated to the text, and hand-sketched worked examples, General Chemistry is still 200 to 300 pages shorter and much less expensive than other two-semester textbooks. Dr. Chang and Dr. Goldsby's concise-but-thorough approach will appeal to efficiency-minded instructors and value-conscious students.

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Chemistry Elsevier

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Textbook Outlines gives all of the outlines, highlights, notes for your textbook with optional online practice tests. Only Cram101 Outlines are Textbook Specific. Cram101 is NOT the Textbook.

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The Periodic Kingdom Cram101

Principles of General Chemistry McGraw-Hill Education

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An unparalleled classic, the sixth edition of Silberberg Chemistry keeps pace with the evolution of student learning. The text maintains unprecedented macroscopic-to-microscopic molecular illustrations, consistent step-by-step worked exercises in every chapter, and extensive range of end-of-chapter problems with engaging applications covering a wide variety of interests, including engineering, medicine, materials, and environmental studies.

Changes have been made to the text and applications throughout to make them more succinct, to the artwork to make it more teachable

and modern, and to the design to make it more modern, simplistic, and open. Features include Three-Level Depictions of Chemical Scenes are the focus of Silberberg's ground-breaking art program, which combines photographs of chemical scenes with an illustrated molecular view and with the equation that symbolically and quantitatively describes that scenario. McGraw-Hill's Connect Chemistry allows teachers to deliver assignments, quizzes, and tests online. Over 2,200 end of chapter problems and additional problems are available to assign. Teachers can edit questions, write new problems, and track student performance.

Student Solutions Manual: Ssm Chemistry Cram101

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Accompanys: 9780073107202 9780073301716 9780073221052 .

Connect 2-Semester Access Card for Principles of General Chemistry McGraw-Hill College

This supplement contains detailed solutions and explanations for all colored problems in the main text.

Nature of Science in General Chemistry Textbooks Principles of General Chemistry

Silberberg ' s Principles of General Chemistry offers students the same authoritative topic coverage as its parent text, Chemistry: The Molecular Nature of Matter and Change. The Principles text allows for succinct coverage of content with minimal emphasis on pedagogic learning aids. This more straightforward approach to

learning appeals to today ' s efficiency-minded, value-conscious instructors and students without sacrificing depth, clarity, or rigor. General Chemistry I as a Second Language Wiley

Research in science education has recognized the importance of history and philosophy of science (HPS). Nature of science (NOS) is considered to be an essential part of HPS with important implications for teaching science. The role played by textbooks in developing students ' informed conceptions of NOS has been a source of considerable interest for science educators. In some parts of the world, textbooks become the curriculum and determine to a great extent what is taught and learned in the classroom. Given this background and interest, this monograph has evaluated NOS in university level general chemistry textbooks published in U.S.A. Most textbooks in this study provided little insight with respect to the nine criteria used for evaluating NOS. Some of the textbooks, however, inevitably refer to HPS and thus provide guidelines for future textbooks. A few of the textbooks go into considerable detail to present the atomic models of Dalton, Thomson, Rutherford, Bohr and wave mechanical to illustrate the tentative nature of scientific theories --- an important NOS aspect. These results lead to the question: Are we teaching science as practiced by scientists? An answer to this question can help us to understand the importance of NOS, by providing students an HPS-based environment, so that they too (just like the scientists) feel the thrill and excitement of discovering new things. This monograph provides students and teachers guidelines for introducing various aspects of NOS, based on historical episodes.

Loose Leaf Version for Principles of General Chemistry with Connect

Access Card McGraw-Hill Education

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Chemistry 2e Cram101

A 'travel guide' to the periodic table, explaining the history, geography and the rules of behaviour in this imagined land. The Periodic Kingdom is a journey of imagination in which Peter Atkins treats the periodic table of elements - the 109 chemical elements in the world, from which everything is made - as a country, a periodic kingdom, each region of which corresponds to an element. Arranged much like a travel guide, the book introduces the reader to the general features of the table, the history of the elements, and the underlying arrangement of the table in terms of the structure and properties of atoms. Atkins sees elements as finely balanced living personalities, with quirks of character and certain, not always outward, dispositions, and the kingdom is thus a land of intellectual satisfaction and infinite delight.

Silberberg, Chemistry (NASTA Reinforced Binding High School) McGraw-Hill Education

Silberberg's Principles of General Chemistry offers students the same authoritative topic coverage as its parent text, Chemistry: The Molecular Nature of Matter and Change. The Principles text allows for succinct coverage of content with minimal emphasis on pedagogic learning aids. This more streamlined approach to learning appeals to today's efficiency-minded, value-conscious instructors and students without sacrificing depth, clarity, or rigor. General Chemistry: The Essential Concepts Academic Internet Pub Incorporated

Emphasises on contemporary applications and an intuitive problem-solving approach that helps students discover the exciting potential of chemical science. This book incorporates fresh applications from the three major areas of modern research: materials, environmental chemistry, and biological science.

Principles of General Chemistry McGraw-Hill Education

This manual contains complete worked-out solutions to all follow-up problems and about half of all the chapter problems. Each chapter of solutions opens with a summary of the text-chapter content and a list of key equations needed to solve the problems.