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This work evolved over thirty combined years of teaching general chemistry to a variety of student demographics. The focus is not to recap or review the theoretical concepts well described in the available texts. Instead, the topics and descriptions in this book make available specific, detailed step-by-step methods and procedures for solving the major types of problems in general chemistry. Explanations, instructional process sequences, solved examples and completely solved practice problems are greatly expanded, containing significantly more detail than can usually be devoted to in a comprehensive text. Many chapters also provide alternative viewpoints as an aid to understanding. Key Features: The authors have included every major topic in the first semester of general chemistry and most major topics from the second semester. Each is written in a specific and detailed step-by-step process for problem solving, whether mathematical or conceptual Each topic has greatly expanded examples and solved practice problems containing significantly more detail than found in comprehensive texts Includes a chapter designed to eliminate confusion concerning acid/base reactions which often persists through working with acid/base equilibrium Many chapters

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This book teaches chemistry at an appropriate level of rigor while removing the confusion and insecurity that impair student success. Students are frequently intimidated by prep chem; Bishop's text shows them how to break the material down and master it. The flexible order of topics allows unit conversions to be covered either early in the course (as is traditionally done) or later, allowing for a much earlier than usual description of elements, compounds, and chemical reactions. The text and superb illustrations provide a solid conceptual framework and address misconceptions. The book helps students to develop strategies for working problems in a series of logical steps. The Examples and Exercises give plenty of confidence-building practice; the end-of-chapter problems test the student's mastery. The system of objectives tells the students exactly what they must learn in each chapter and where to find it. Cracking the GRE Chemistry Subject Test W. W. Norton Writing chemical reactions in general and inorganic chemistry is not a trivial task. However, writing reactions for chemical processes correctly is a clear indicator of proficiency and competence in a subject. Unfortunately, very few students grasp the concept of the correct writing of chemical reactions quickly, and so are unable to move through topics of general, analytical, and inorganic chemistry freely. Because the ability to write and balance different types of chemical reactions is a fundamental issue, this becomes a key question of chemical literacy. The successful writing of chemical reactions includes two components: the prediction of products of these reactions and their possible variations, and balancing these reactions providing a material balance between starting compounds and reactions 'products. This book explores that element of the teaching of the fundamentals of chemical literacy: writing complete equations of chemical reactions and balancing them. It contains 49 figures, 22 schemes and 12 tables, and 93 problems (with answers). This book will be very useful for high school students interested in chemical sciences, higher education teachers, students in colleges and universities majoring in chemistry and biochemistry, and chemistry professional working in industry. It also contains information about properties of the most common elements and applications of a variety of their chemical compounds.

Merrill Chemistry John Wiley & Sons

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you. I believe that this Section 3 would greatly help students with levels varying from high school to advanced university classes. Student Study Guide Fundamentals of Chemistry by David E. Goldberg John Wiley & Sons

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