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Restoring Balance LAB Pre-Lab Questions 1. The solubility of iodine increases because the position of equilibrium has shifted towards the triiodide side of the reaction. Addition of the reactants causes an increase in the product which is directly related to the solubility of iodine in water in this case.

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Restoring Balance Lab

1. Prepare beakers with hot and iced water for Part B: Fill a 250-mL beaker half way with tap water. Place it on a hot plate and heat to 65-70 degrees. In another 250-mL beaker, add water and ice for an ice-water bath. 2. Obtain a 50-mL beaker and pour 20 mL of a 1% solution of cobalt chloride in alcohol. 3.

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1. Place test tube B in a hot water bath at 65-70 C for 3 minutes. Record the initial and final color of the solution. 2. Place test tube C in an ice bath at 0-5 C for 5 minutes. Record the initial and final color of the solution.

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