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51. Write a point of distinction between a metallic solid and an ionic solid other than metallic lustre. Answer: Metallic solids can conduct electricity due to movement of electrons in solid state, whereas ionic solids conduct electricity in molten state or in aqueous solution only due to the movement of ions. 52.

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A solid has 'bcc' structure. If the distance of nearest approach between two atoms is 1.73 Å, the edge length of the cell is :-(1) 314.20 pm (2) 216 pm (3) 200 pm (4) 1.41 pm. Jee-Main (online)-12

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