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## Solubility Aqueous Solutions

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The Solubility of the Phosphates of Calcium  
in Aqueous Solutions of Sulfur Dioxide  
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Excerpt from The Action of Water and Aqueous Solutions Upon Soil Carbonates  
The solubility in water of carbon dioxide, like all other gases, is greater at the lower temperatures than at the higher temperatures. With one or possibly two known exceptions, the solubility in aqueous solutions is decreased by increasing quantities of the material in solution. Thus, the solubility of carbon dioxide in water is decreased either by increasing the temperature or by the addition of some material, such as sodium chloride or other salts. The results of the work recorded in the literature have been assembled and are given in the following tables. The results are given in the same form as they have been recorded in the original papers. For instance, the solubility of carbon dioxide In

water at  $10^{\circ}\text{C}$ . Has been given by Bunsen as This means that one cubic centimeter of water at  $10^{\circ}$  will dissolve the quantity of carbon dioxide occupying cubic centimeters at  $0^{\circ}$  and 760mm. All the gaseous volumes are reduced to  $0^{\circ}$  and 760 mm. Pressure. In this way comparisons may be made between the solubility of the gas in Solvents at different temperatures and also in different Solutions. About the Publisher  
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<u>A New Method for Determining the Solubility of Salts in Aqueous Solutions at Elevated Temperatures</u>	<u>The Solubility of Chlorine in Aqueous Solutions of Chlorides and the Free Energy of Trichloride Ion, by M.S. Sherrill and E.F. Izard</u>
<i>Solubility of Dimethylmercury in Aqueous Solutions</i>	<b>Factors Affecting Solubility of Calcium Lactate in Aqueous Solutions</b>
<u>Solubility of Carbon Dioxide in Aqueous Solutions of Alcohols</u>	<i>The Solubility of Iodine in Aqueous Solutions of Bromides of Potassium and Sodium</i>
<b>Solubility and Adsorption of Antimony (III) in Aqueous Solutions of Varying PH</b>	<b>The Solubility of Aniline in Aqueous Solutions of Its Hydrochloride</b>
<b>A Study of the Solubility of Certain Organic Substances in Aqueous Solutions</b>	<u>Solubility of Sphalerite in Aqueous Solutions of Sodium Chloride at High Temperatures</u>
<b>Solubility of Certain Electrolytes in Aqueous Solutions of More Soluble Electrolytes</b>	<i>The Solubility of Carbon Dioxide in Aqueous Solutions Containing Alcohols and Sugars</i>
<b>Solubility of Carbonates in Aqueous Solutions</b>	
<b>Solubility of Oxygen in Aqueous Solutions of LiF and LiClO<sub>4</sub></b>	
<u>The Solubility of Magnetite in Water and in Aqueous Solutions of Acid and Alkali</u>	