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4 Problems and Solutions in Atomic, Nuclear and Particle Physics Solution: The Bohr radius of the hydrogen atom and the Compton wavelength of electron are given by $a = \frac{h^2}{m_e e^2}$ and $\lambda_c = \frac{h}{m_e c}$ respectively. Hence $a = \lambda_c =$

$\frac{1}{2} \left(\frac{e^2}{c} \right) - \frac{1}{2} \left(\frac{e^2}{c} \right) = 22$, where $\frac{e^2}{c}$ is the fine-structure constant. Hence the answer is (a). 1004

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