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Thermodynamics of Materials 5th edition by David R Gaskell Chapter 11 Homework Solution Tips

11.1 This is a reaction extent problem. Note: for gases: volume percent = mole percent = pressure percent Get K from  $\Delta G^\circ$

11.2 This is a reaction extent problem for the reaction  $\text{SO}_2 + 0.5 \text{O}_2$ .  
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...  
Thermodynamics began with the study of heat and work effects and relations between heat and work. Some early thermodynamics problems were for very practical problems. For example, in a steam engine heat is supplied to water to create steam. The steam

is then used to turn an engine which does work. Finally, the water is exhausted to

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Thermodynamics of solutions. Enthalpy of mixing 4:56 Entropy of Mixing 24:14 Gibb's Energy of Mixing (The Regular Solution Model) 39:56 ... Part 1 of lecture 5.

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 Needed for Problems ü  
 Heat Capacities The  
 generic heat capacity  $C_p$   
 $= a + b T$   
 $\frac{1}{10^3} + c \frac{1}{10^5} \frac{1}{T^2}$  ; The  
 heat capacities of various  
 elements and compounds  
 are  $C_p = a + b T + c \frac{1}{T^2}$   
 $21.30, b = 8.54, c = 1.51 \times 10^{-5}$ ;  
 $C_p = a + b T + c \frac{1}{T^2}$  ...  
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Materials (6th ed.)  
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 Materials, Fifth Edition  
 (David R. Gaskell) Problem  
 13.13: UC2 can be  
 equilibrated with UC and C  
 at high temperature and can  
 be equilibrated with U2C3  
 and C at lower  
 temperatures. Calculate the  
 maximum and minimum  
 temperatures at which UC2  
 can exist.  
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 6 Heat Capacity,  
 Enthalpy, Entropy, and  
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