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colloidal solution will be 1 - 1000 nm in diameter.

(3). Suspension: The size of particles in a suspension will be greater than 1000 nm. Suspension is a heterogenous mixture of two or more substances.

Hence, this is another difference between solution and colloid. Furthermore, one other difference between solution and colloid is that the colloidal are either opaque or translucent, but solutions are transparent. Summary - Solution vs Colloid. Both solutions and colloids are mixtures of two or more substances. The key difference between solution and colloid is that the particles in a colloid are often bigger than the solute particles in a solution. Reference: 1.