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ous systems involving salt-type minerals are examined in this work. Stability relations have been discussed on the basis of bulk equilibria.

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EQUILIBRIA PART 1: CARBONATES.

Carbonic acid and the carbonate minerals provide another good illustration of the use of equilibrium reasoning in geochemistry. Interactions among these compounds determine the conditions under which limestones and dolomites are formed or dissolved, and likewise the conditions of formation of carbonate minerals as cements in soils and sandstones and as vein fillings.