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Short Cut. For compounds containing only C, H, O, N, S, and halogens, the following steps permit a quick and simple calculation of the number of double bond equivalents: 1. O and divalent S are deleted from the molecular formula 2. Halogens are replaced by hydrogen 3. Trivalent N is replaced by CH 4.

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Spectrometric Identification Of Organic Compounds

- (CH<sub>2</sub>=CH<sub>2</sub>) - (H<sub>2</sub>O) 1 2 Triplet Plus Appendix B.45

for ClCH<sub>2</sub> plus Table B.1 value of 1.6 1 Doublet 9.50

for HC=O 3.1(m) 1 and 2 is a spin system.80

extrapolation Calc is based on Chart A1 3. Pople notation

is A2X.1 to 3.1\* 2 Appendix D. Answers for Student

Exercise 3. 2 protons are enantiotopic. Table B.3 (m) 2 1

Integration 1 2 .