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The methods in the 22nd edition (as in previous editions) are believed to be the best available, generally accepted procedures for analyzing water, wastewater, and related materials. They represent the recommendations of specialists, ratified by a large number of analysts and others of more general expertise, and as such are truly consensus standards, offering a valid and recognized basis for control

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of Water and Wastewater ERRATA Despite the best efforts of those involved in the manufacture of the 22nd Edition of Standard Methods, the following errors have occurred in the text: 1. In Section 1020B.1 (p. 1-8), the second equation should be “ Lower control limit = Mean Standard Methods Standard Methods for the Examination of Water and Wastewater represents the best current practice of American water analysts. This comprehensive reference covers all aspects of water and wastewater analysis techniques. Standard Methods where: $\text{pH} \uparrow = \log 10 a \text{H}^+$ and. pOH

= log 10 a OH – ..
 Equation 2 states
 that as pH increases
 pOH decreases
 correspondingly and
 vice versa because
 pK w is constant for
 a given temperature.
 At 25 ° C, pH 7.0 is
 neutral, the activities
 of the hydrogen and
 hydroxyl ions are
 equal, and each
 corresponds to an
 approximate activity
 of 10 – 7 moles/L.
 The neutral point is
 temperature-
 dependent and is ...
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Williams

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For methods where bias is applicable (see Table 2020:1), run a laboratory-fortified blank (LFB) (2020B.2e), performance evaluation sample, or standard with a known or otherwise certifiable

concentration at least four times and compare results to the limits listed in the method or those established by the laboratory. If no limit is specified, use the following procedure to establish limits:
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