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Atoms, Molecules & Stoichiometry | Multiple Choice | Mark ...  
Stoichiometry and the Mole The relative atomic masses of sodium, oxygen and hydrogen are 23, 16 and 1 respectively. How many grams are there in 0.5 mole of sodium hydroxide (NaOH)? 38 grams  
Chemistry 103 Assignment No. 9 Reaction and Stoichiometry ...  
Multiple Choice and Short Answer. Problem One. In the oxidation of ethane: 2 C 2 H 6 + 7 O 2 → 4 CO 2 + 6 H 2 O. how many moles of O 2 are required to react with 1 mole of ethane? a) 7 moles b) 2 moles c) 7/2 moles. Incorrect. This is the coefficient for O 2, but the mole ratio of ethane to O 2 is 7/2. Incorrect.  
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Practice: Stoichiometry questions. This is the currently selected item. Stoichiometry article. Stoichiometry and empirical formulae. Empirical formula from mass composition edited. Molecular and empirical formulas. The mole and Avogadro's number. Stoichiometry example problem 1. Stoichiometry.  
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The following section consists of Chemistry Multiple Choice questions on Stoichiometry. Take the Quiz for competitions and exams.  
**Stoichiometry Multiple Choice Questions And Answers**  
Multiple Choice Questions (MCQ) and Answers on Stoichiometry. Question 1 : The weight fraction of methanol in an aqueous solution is 0.64. The mole fraction of methanol X M satisfies  
**Stoichiometry questions (practice) | Khan Academy**  
Stoichiometry Multiple Choice Questions and Answers - Set 01 Chemical MCQ Edit Practice Test: Question Set - 01. 1. In the reaction, Ca + 2H 2 O = Ca(OH) 2 + H 2; what volume (c.c.) of hydrogen at STP would be liberated, when 8 gm of calcium reacts with excess water ? (Atomic weight of calcium = 40).  
[Ch 3 Stoichiometry Multiple Choice \[klzz6oekx7lg\]](#)  
Reaction and Stoichiometry MULTIPLE CHOICE QUESTIONS Select the one best answer for each question. A. If 1.00 g of an unknown molecular compound contains 4.55 × 1021 molecules, what is its molar mass? 1. 44.0 g/mol 2. 66.4 g/mol 3. 72.1 g/mol 4. 98.1 g/mol 5. 132 g/mol B. What is the mass percent of each element in dichloromethane, CH 2Cl 2? 1.  
**Multiple Choice and Short Answer - Wired Chemist**  
Balance: CH4 + 2O2 → CO2 + 2H2O Then do some stoichiometry using “easy math” 16 g of methane (MM = 16) is 1 mole and 1 mole of methane will produce 1 mole of CO2 = 44 g, and 2 moles of H2O which is 36 g for a total of 80 g Balance: C3H8 + 5O2 → 3CO2 + 4H2O Balance: 2KClO3 → 2KCl + 3O2 In multiple choice questions without a calculator, you must look for the “easy math” ? You will be most successful at this if you put all the numbers in the dimensional analysis on the page and ...  
[Senior Chemistry - ScienceQuiz.net](#)  
Consider the following balanced equation. C 12 H 22 O 11 + 3O 2 → 2H 3 C 6 H 5 O 7 + 3H 2 O Determine the mass of citric acid (H 3 C 6 H 5 O7) produced when 2.5 mol C 12 H 22 O 11 is used.  
*AP Chemistry: Stoichiometry – Multiple Choice Answers*  
Stoichiometry Multiple Choice question? According to the reaction 2AI + 3H3SO4 --> 3H2 + AI2 (SO4)3, the total number of moles of H2SO4 needed to react completely with 5.0 mol of AI is. I got 3,...  
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AP Chemistry: Stoichiometry – Multiple Choice Answers 44. What number of moles of O 2 is needed to produce 14.2 grams of P 4O 10 from P? (Molar Mass P 4O 10 = 284) (A) 0.0500 mole (B) 0.0625 mole (C) 0.125 mole (D) 0.250 mole (E) 0.500 mole 4 P + 5 O 2 → P 4O 10 >> 14.2 g P 4O 10 x 1 mol P 4O 10 x 5 mol O 2 \_\_\_\_ = 0.250 mol 284 g P 4O  
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