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Practice balancing chemical equations by changing the coefficients of reactants and products. As the equation is manipulated, the amount of each element is shown as individual atoms, histograms, or numerically. Molar masses of reactants and products can also be calculated and balanced to demonstrate conservation of mass.

Student exploration balancing

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This process is summarized by a chemical equation. In the Balancing Chemical Equations Gizmo™, look at the floating molecules below the initial reaction: $H_2O_2 \rightarrow H_2O$. 1. How many atoms are in a hydrogen molecule (H_2)? $2 + 2$. How many atoms are in an oxygen molecule (O_2)? 2×3 . How many hydrogen and oxygen atoms are in a water molecule (H_2O)? 3×4 .

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Adjust coefficients to balance a chemical equation. Classify chemical equations as combination (synthesis), decomposition, single replacement, or double replacement.

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Student Exploration: Chemical Equations Combustion – a fuel reacts with oxygen to release carbon dioxide, water, and heat. $CH_4 + O_2 \rightarrow CO_2 + H_2O$ $C_3H_8 + O_2 \rightarrow CO_2 + H_2O$ $C_6H_{12}O_6 + O_2 \rightarrow CO_2 + H_2O$ Single replacement – an element replaces another element in a compound. $KCl + F_2 \rightarrow KF + Cl_2$ $Mg + HCl \rightarrow MgCl_2 + H_2$,

$\text{Cu} + \text{AgNO}_3 \rightarrow \text{Cu}(\text{NO}_3)_2 + \text{Ag}$ Double replacement — two compounds switch parts with one another. $\text{AgNO}_3 + \text{K}_2\text{SO}_4 \rightarrow \text{Ag}_2\text{SO}_4 + \text{KNO}_3$ $\text{Mg}(\text{OH})_2 + \text{HCl} \rightarrow \text{MgCl}_2 + \text{H}_2\text{O}$...

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answers. . Writing chemical equations, balancing chemical reactions . Balancing Chemical Equations Gizmo Assessment Answers Therefore, a balanced chemical equation will show the same number of each type of atom on each side of the equation. To set up an equation in the Chemical Equations Gizmo, type the chemical... Student

Exploration ...

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Student Exploration: Balancing Chemical Equations. DEFINE Vocabulary: coefficient. Combination- a chemical reaction when two reactions combine to make one. Compound- a substance consisting of two or more elements . Decomposition- the separation of a substance into a simpler form . Double replacement- multiple elements that are getting swapped

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2018 Name: Simon Alam Date: 12/7/2020

_____ Student Exploration: Balancing Chemical Equations Vocabulary: coefficient, combustion, compound, decomposition, double replacement, element, molecule, product, reactant, single replacement, subscript, synthesis Prior Knowledge Questions (Do these BEFORE using the Gizmo.) The scouts are making s ' mores out of toasted marshmallows, chocolate, and ...

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