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Practice balancing chemical equations by changing the coefficients of reactants and products. As the equation is manipulated, the amount of each element is shown as individual atoms, histograms, or numerically. Molar masses of reactants and products can also be calculated and balanced to demonstrate conservation of mass.

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This process is summarized by a chemical equation. In the Balancing Chemical Equations Gizmo™, look at the floating molecules below the initial reaction: $H_2 + O_2 \rightarrow H_2O$. 1. How many atoms are in a hydrogen molecule (H_2)? 2 + 2. How many atoms are in an oxygen molecule (O_2)? 2 + 2. How many hydrogen and oxygen atoms are in a water molecule (H_2O)? 3 + 1. Chemical Changes Gizmo Answer Key - YouTube

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answers. . Writing chemical equations, balancing chemical reactions . Balancing Chemical Equations Gizmo Assessment Answers Therefore, a balanced chemical equation will show the same number of each type of atom on each side of the equation. To set up an equation in the Chemical Equations Gizmo, type the chemical... Student Exploration ...

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Student Exploration: Balancing Chemical Equations Learning Objectives

Students will... Adjust coefficients to balance a chemical equation.

Classify chemical equations as combination (synthesis), decomposition, single

replacement, or double replacement.

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Student Exploration: Balancing Chemical Equations. DEFINE Vocabulary:

coefficient. In the Balancing Chemical Equations Gizmo™, look at the floating

molecules below the initial reaction: $H_2 + O_2 \rightarrow H_2O$. When you are done, turn on

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Student Exploration: Chemical Equations Combustion – a fuel reacts with oxygen to

release carbon dioxide, water, and heat. $CH_4 + O_2 \rightarrow CO_2 + H_2O$ $C_3H_8 + O_2$

$CO_2 + H_2O$ $C_6H_{12}O_6 + O_2 \rightarrow CO_2 + H_2O$ Single replacement – an element

replaces another element in a compound. $KCl + F_2 \rightarrow KF + Cl_2$ $Mg + HCl \rightarrow MgCl_2$

$+ H_2$, $Cu + AgNO_3 \rightarrow Cu(NO_3)_2 + Ag$ Double replacement – two compounds

switch parts with one another. $AgNO_3 + K_2SO_4 \rightarrow Ag_2SO_4 + KNO_3$ $Mg(OH)_2 + HCl \rightarrow MgCl_2$...

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Name: Arianna Noble Date: 2/26/2017 Student Exploration: Chemical Equations

Vocabulary: Avogadro's number, chemical equation,

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minutes. Our state-specific web-based samples and complete guidelines remove human-

prone faults. ... Student Exploration Chemical Equations Answer Key.

Balancing Chemical Equations Gizmo Worksheet Answers ...

2 Gizmo Warm-up In a chemical reaction, reactants interact to form products. This

process is summarized by a chemical equation. In the Balancing Chemical Equations

Gizmo, look at the floating molecules below the initial reaction: $H_2 + O_2 \rightarrow H_2O$.

How many atoms are in a hydrogen molecule (H_2)? 2

O. 2

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2018 Name: Simon Alam Date: 12/7/2020 ____ Student Exploration: Balancing Chemical

Equations Vocabulary: coefficient, combustion, compound, decomposition, double

replacement, element, molecule, product, reactant, single replacement, subscript, synthesis

Prior Knowledge Questions (Do these BEFORE using the Gizmo.) The scouts are making

s'mores out of toasted marshmallows, chocolate, and ...

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Student Exploration: Balancing Chemical Equations. DEFINE Vocabulary: coefficient.

Combination- a chemical reaction when two reactions combine to make one. Compound-

a substance consisting of two or more elements . Decomposition- the separation of a

substance into a simpler form . Double replacement- multiple elements that are getting

swapped

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Balancing Chemical Equations Gizmo, look at the floating molecules below the

initial reaction: Balancing chemical equations gizmo assessment answers. . .

Check your answers and then write the balanced equations. Assessment:

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chemical equations, balancing chemical reactions .