

## Student Exploration Stoichiometry Gizmo Answers

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Read Free Student Exploration Stoichiometry Gizmo Answer Key try your hand at figuring out the amounts of reactants and products that take part in a chemical reaction. To begin, check that this equation is shown:  $\text{Fe}_2\text{O}_3 + 3\text{CO} \rightarrow 2\text{Fe} + 3\text{CO}_2$ . Look at the coefficients (such as the "3" in  $3\text{CO}$ ) in front of each substance in the equation.

Student Exploration Stoichiometry Answer Key

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Student Exploration Diffusion Worksheets - Kiddy Math

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#### **Student Exploration Stoichiometry Gizmo Answers**

URL: [www.explorelearning.com](http://www.explorelearning.com) LOGIN: gablechem PASSWORD: chem317. In the SEARCH bar type: stoichiometry ... select the 1st gizmo titled STOICHIOMETRY that appears. The first Gizmo shows this equation:  $Fe_2O_3(s) + 3CO(g) \rightarrow 2Fe(s) + 3CO_2(g)$  There is a Question to solve under the equation. Here are the features that will help you solve each problem:

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Stoichiometry Gizmo™ allows you to try your hand at figuring out...

The correct answer, of course, is E. In chemistry, the mole (mol) is defined as an amount of a. <https://www.rhnet.org/site/handlers/filedownload.ashx?moduleinstanceid=3324&dataid=107450&FileName=Gizmo%20%20Stoichiometry.pdf> read more

#### **Gizmo: Stoichiometry**

Stoichiometry. Launch Gizmo. Solve problems in chemistry using dimensional analysis. Select appropriate tiles so that units in the question are converted into units of the answer. Tiles can be flipped, and answers can be calculated once the appropriate unit conversions have been applied. Video Preview Launch Gizmo.

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2019 Name: \_\_\_\_\_ Date: \_\_\_\_\_ Student Exploration: Stoichiometry Vocabulary: Avogadro's number, balanced equation, cancel, coefficient, conversion factor, dimensional analysis, molar mass, mole, molecular mass, stoichiometry Prior Knowledge Questions (Do these BEFORE using the Gizmo.) 1. A 250 mL glass of orange juice contains 22 grams of sugar.

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reaction. to begin, check that this equation is shown:  $\text{Fe}_2\text{O}_3 + 3\text{CO}$   
 $2\text{Fe} + 3\text{CO}_2$  1. look at the coefficients (such as the Page 11/30