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something about a specific situation. There are
many kinds of graphs. One of the most common is
the bar graph.

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Matter and Change • Chapter 21 123 In your
textbook, read about calculating cell
electrochemical potential. Use the table of
standard reduction potentials below to answer
the following questions. 20. Suppose you have
two voltaic cells whose half-cells are
represented by the following

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Science: Geology, the Environment, and the
Universe1 SECTION 1.1 Earth Science In your
textbook, read about the scope of Earth science. Use
the terms below to identify of the major area of Earth
science that studies each subject. Each term can be
used more than once. astronomy meteorology
geology ...

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Matter and Change • Chapter 3 13

Matter—Properties and Changes
Matter—Properties
and Changes Section 3.1 Properties of Matter
In your textbook, read about physical properties and
chemical properties of matter. Use each of the
terms below just once to complete the passage.

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Study Guide for Content Mastery Chemistry:

Matter and Change • Chapter 2 11

In your textbook, read about significant figures. Use

each of the terms below just once to complete
the statements. 7. The digits that are reported in
an answer are called . 8. The numeral 9.66 has
three significant figures, two known figures and
one figure.

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Matter—Properties and Changes

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Chemistry: Matter and Change • Chapter 10

59 Section 10.3 Reactions in Aqueous

Solutions In your textbook, read about

aqueous solutions, reactions that form

precipitates, reactions that form water, and

reactions that form gases.

Chapter 21 Study Guide For Content Mastery Answers ...

Study Guide for Content Mastery Chemistry:

Matter and Change • Chapter 2 7 Data

AnalysisData Analysis Section 2.1 Units of

Measurement In your textbook, read about SI units.

Complete the following table. In your textbook,

read about base units and derived units. For each SI unit in Column A, write the letter of the matching item from Column B.

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This Study Guide for Content Mastery for

Earth Science: Geology, the. Environment, and

the. Each textbook chapter has six study guide

pages of questions and activities for you. Circle

the letter of the choice that best completes the

statement or answers the question.

Chemistry: Matter and Change; Study Guide for Content ...

Study Guide for Content Mastery Answer

Key Chemistry: Matter and Change T195

Name Date Class 76 Chemistry: Matter and

Change • Chapter 13 Study Guide for

Content Mastery Section 13.3 Liquids and

Solids In your textbook, read about liquids

and solids. In the space at the left, write true if the statement is true; if the statement is false,

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you read the text. The study guide pages are divided into sections that match those in your text. You will find that the directions in the Study Guide for Content Mastery are simply stated and easy to follow. Sometimes you will be asked to answer questions. Other times, you will be asked to interpret a diagram or complete a table. By completing the study guide, you will gain a better understanding

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Chemistry: Matter and Change T207 Name
Date Class 100 Chemistry: Matter and Change
• Chapter 17 Study Guide for Content Mastery
Section 17.3 Reaction Rate Laws In your

textbook, read about reaction rate laws and determining reaction order. Use each of the terms below to complete the statements ...

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Class STUDY GUIDE FOR CONTENT Section
4.4 Changes to the Nucleus—Nuclear Reactions In
your textbook, read about radioactivity. For each
item in Column A, write the letter of the matching
item in Column B. Column A 1. The rays and
particles that are emitted by a radioactive material
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T182 Chemistry: Matter and Change Name

Date Class Name Date Class CHAPTER 9

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CHAPTER 9 STUDY GUIDE FOR

CONTENT MASTERY Covalent Bonding

Section 9.2 Naming Molecules In your textbook, read about how binary compounds and acids are named from their formulas.

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Environment and the Universe- Study

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(9780078245664): McGraw-Hill: Books

STUDY GUIDE FOR CONTENT

MASTERY Section 6.3 Periodic Trends In

your textbook, read about atomic radius and

ionic radius. Circle the letter of the choice

that best completes the statement or answers

the question. 1. Atomic radii cannot be

measured directly because the electron

cloud surrounding the nucleus does not

have a clearly defined c. outer edge.