

# Study Guide For Content Mastery Chapter 13

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Study Guide for Content Mastery . Name Date Class STUDY GUIDE FOR CONTENT Section 4.4 Changes to the Nucleus—Nuclear Reactions In your textbook, read about radioactivity. For each item in Column A, write the letter of the mmching item in Column B. Column A 1. The rays and particles that are emitted by a radioactive material

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Study Guide for Content Mastery Chemistry: Matter and Change • Chapter 3 13  
Matter—Properties and ChangesMatter—Properties and Changes Section 3.1 Properties of Matter In your textbook, read about physical properties and chemical properties of matter. Use each of the terms below just once to complete the passage.

T182 Chemistry: Matter and Change Name Date Class Name Date Class CHAPTER 9 STUDY GUIDE FOR CONTENT MASTERY CHAPTER 9 STUDY GUIDE FOR CONTENT MASTERY Covalent Bonding Section 9.2 Naming Molecules In your textbook, read about how binary compounds and acids are named from their formulas.  
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This Study Guide for Content Mastery for Earth Science: Geology, the. Environment, and the. Each textbook chapter has six study guide pages of questions and activities for you. Circle the letter of the choice that best completes the statement or answers the question.

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Study Guide for Content Mastery Chemistry: Matter and Change • Chapter 2 11 In your textbook, read about significant figures. Use each of the terms below just once to complete the statements. 7. The digits that are reported in an answer are called . 8. The numeral 9.66 has three significant figures, two known figures and one figure.  
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Study Guide for Content Mastery Chemistry: Matter and Change • Chapter 2 7 Data AnalysisData Analysis Section 2.1 Units of Measurement In your textbook, read about SI units. Complete the following table. In your textbook, read about base units and derived units. For each SI unit in Column A, write the letter of the matching item from Column B.  
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STUDY GUIDE FOR CONTENT MASTERY Section 6.3 Periodic Trends In your textbook, read about atomic radius and ionic radius. Circle the letter of the choice that best completes the statement or answers the question. 1.

Atomic radii cannot be measured directly because the electron cloud surrounding the nucleus does not have a clearly defined c. outer edge.  
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