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1) An atom has hardly any empty space, and the nucleus is positive charge 2) An atom has hardly any empty space, and the nucleus is negative charge 3) An atom is mainly empty space, and the nucleus has a positive charge 4) An atom is mainly empty space, and the nucleus has a negative charge. Practice 5.

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16 From “Surviving Chemistry: Workbook” e3chemistry.com adius (pm) Worksheet 7: Trends on the Periodic Table Topic 2 Set B: Data and Graphing for Period 2 Elements Using Reference Table S, Complete the table below for Period 2 Elements. Once done, plot and graph the data on the graphing grids to observe trends of the four properties. ...

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