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(b) given that  $s = 9$  mol/kg for pure LiCl in H<sub>2</sub>O. Using the Meissner model for predictions for both pure and mixed electrolyte.  $K_{sp} = 9 \times 10^{-16}$  for pure LiCl at 25 °C. Iterating  $s^* = 2.4$  and 3H for mixed LiCl and HCl at 25 °C  
(c) Difficult to say without more information on how much Na<sub>2</sub>SO<sub>4</sub> influences for LiCl. For example is the effect on large enough to

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