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solution, devoid of conceptual or technical errors. 2.5: solution is conceptually correct, but contains minor mathematical errors (signs, numerical

factors, etc.)

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(alnZ (5.1) When a system is in thermal and diffusive equilibrium with a reservoir, the grand partition function is  $Z = \sum e^{-\epsilon_i} = \sum e^{-\epsilon_i - \mu N_i} = e^{-\beta \Omega}$ , where  $\epsilon_i$

represents microstates in the system. Prove that the average number of particles in the system is  $\bar{n} = - \left( \frac{\partial \ln Z}{\partial \mu} \right) = \beta \left( \frac{\partial \Omega}{\partial \mu} \right)$  (5.2) An absorption plate has N absorbing

sites, each of which can absorb at most one ideal gas ...

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in Schroeder),  $DH H_2O = 285.83 \text{ kJ mol}^{-1}$  and  $DH = 890.36 \text{ kJ mol}^{-1}$  (12) This time, the final volume is 1/3 of the initial volume, since the 2

moles of water has condensed out as liquid with negligible volume compared to the gases. Thus the atmosphere does work  $PdV = RT \ln \left( \frac{V_f}{V_i} \right) =$

$8.31 \text{ J K}^{-1} (298 \text{ K}) \ln \left( \frac{1}{3} \right) = 4.953 \text{ kJ}$  (15)

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