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~~Thermodynamic Properties of Ideal Solution~~
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~~Lec # 5 II Ideal and Non Ideal Solution II Dr Rizwana~~
~~IDEAL AND NON IDEAL SOLUTIONS~~
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Thermodynamics Ideal Solution 7 Edition and obtain the equation of state for an ideal gas. Solution. $1/2 \int_{V_1}^{V_2} \frac{1}{V} dV = \ln \frac{V_2}{V_1}$ $\int_{T_1}^{T_2} \frac{1}{T} dT = \ln \frac{T_2}{T_1}$ $\int_{P_1}^{P_2} \frac{1}{P} dP = \ln \frac{P_2}{P_1}$ $\int_{T_1}^{T_2} \frac{1}{T} dT = \ln \frac{T_2}{T_1}$ $\int_{P_1}^{P_2} \frac{1}{P} dP = \ln \frac{P_2}{P_1}$ $\int_{T_1}^{T_2} \frac{1}{T} dT = \ln \frac{T_2}{T_1}$ $\int_{P_1}^{P_2} \frac{1}{P} dP = \ln \frac{P_2}{P_1}$ The constant is nR for n moles of the ideal gas.

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