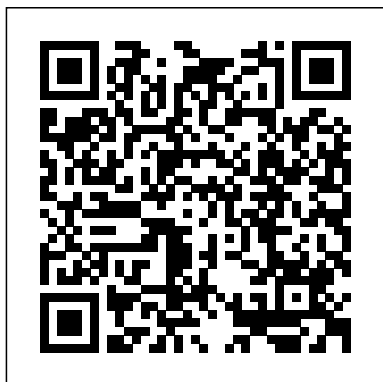

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 The hydraulic lift in a car repair shop is to
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 maintained in the reservoir is to be
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 piston of the lift is negligible.
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 is the sixth chapter in the NCERT class 11
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 branch of science that deals with the

relationship between heat and other forms of energy. A part of the universe where observations are made is called system.

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Answer: (i) First law of thermodynamics: It states that energy can neither be created nor be destroyed. The energy of an isolated system is constant. The energy of an isolated system is constant. $\Delta u = q + w$

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Given that the free energy of formation of liquid water is -237 kJ / mol , calculate the potential for the formation of hydrogen and oxygen from water. To solve this problem we must first calculate ΔG for the reaction, which is $-2 (-237 \text{ kJ / mol}) = 474 \text{ kJ / mol}$. Knowing that $\Delta G = -nFE^\circ$ and $n = 4$, we calculate the potential is -1.23 V .

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Solution Manual, Yunus Cengel, Thermodynamics, 8th edition ...

A thermodynamic system works by sending a very cold liquid refrigerant around a panel on the outside of a property. The temperature of the liquid is approximately -22°C when it enters

the panel. As the liquid flows through the veins in the panel it absorbs heat energy from the air surrounding the panel.

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