

Titration Gizmo Assessment Answers

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answers to titration gizmo - thepopculturecompanyintroduction & titration gizmo answer key teacher guide measure: a titration can be used to determine the concentration of an acid or base by measuring the amount of a solution with a known concentration, called the titrant, which reacts completely with a solution of unknown concentration, called the analyte.

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Measure the quantity of a known solution needed to neutralize an acid or base of unknown concentration. Use this information to calculate the unknown concentration. A variety of indicators can be used to show the pH of the solution.

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questions 4 and 5. Check your answers with the Gizmo. (Because dissociation is not always complete, your answers may vary slightly from values in the Gizmo.) $[H_2SO_4] = 0.108 M$ $pH H_2SO_4 = 0.967$ $[HBr] = 0.210 M$ $pH HBr = 0.678$ 7. Apply: For a strong base, the concentration of hydroxide ions $[OH^-]$ is roughly estimated to

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Ph Analysis Gizmo Answer Key Acidic substances have a pH below 7, while alkaline substances (bases) have a pH above 7. Pure water has a pH of 7 and is considered neutral. The pH Analysis Gizmo™ allows you to find the pH of a variety of liquids. In the Gizmo, check that the Substance in the tube is Ammonia, and click Test.