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point in titration at which the amount of titrant added is just enough to completely neutralize the analyte solution. At the equivalence point in an acid-base titration, moles of base = moles of acid and the solution only contains salt and water.

Titration - for Gr 11 Chemistry, University Prep . I adapted the student worksheets to meet the expectations of the Ontario Curriculum. I removed the titrat... (more) ion curves and set it up as a "discovery" exploration. Students practice a titration before completing the real thing and incorporate stoichiometry for neutralization reactions.

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In the Titration Gizmo™, you will use indicators to show how acids are neutralized by bases, and vice versa. To begin, check that 1.00 M NaOH is selected for the Burette, Mystery HBr is selected for the Flask, and Bromthymol blue is selected for the Indicator. 1. Look at the ask. What is the color of the bromthymol blue indicator? \_ 2.

#### Gizmo Orbital Motion Answer Key - Legacy

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