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The borax chemical formula is also commonly written in a more compact notation as $\text{Na}_2\text{B}_4\text{O}_7 \cdot 10\text{H}_2\text{O}$. Sodium borate can be obtained in high purity and so can be used to make a standard solution in titrimetric analysis. A number of metal borates are known. They are produced by treating boric acid or boron oxides with metal oxides.

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